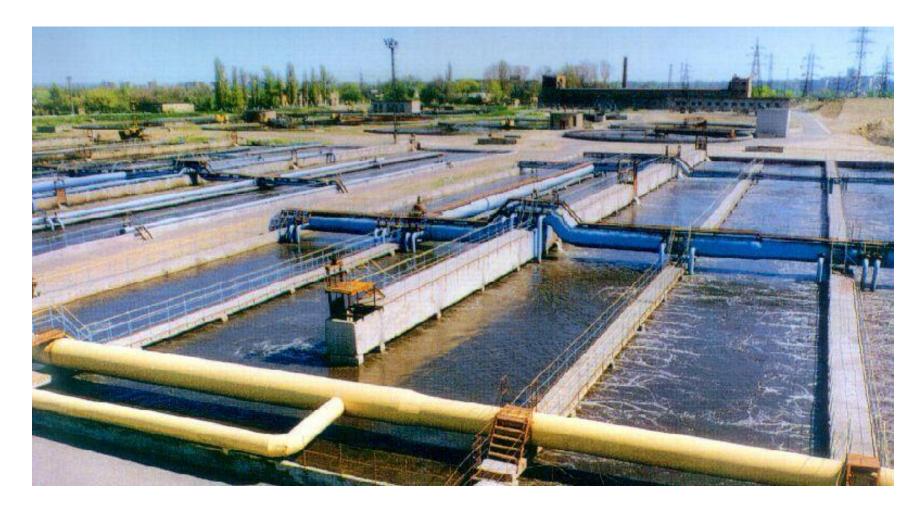




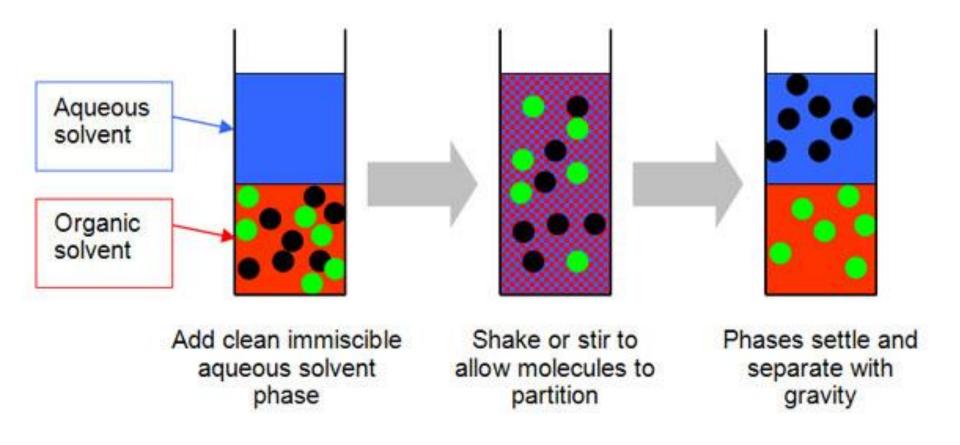
STUDY OF PROSPECTIVE COMPLEXES OF F-ELEMENTS WITH ANILIDES TO CLEAN INDUSTRIAL WASTE WATERS

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The use of f-elements from wastes separated during the purification of industrial waste water



A selective extraction

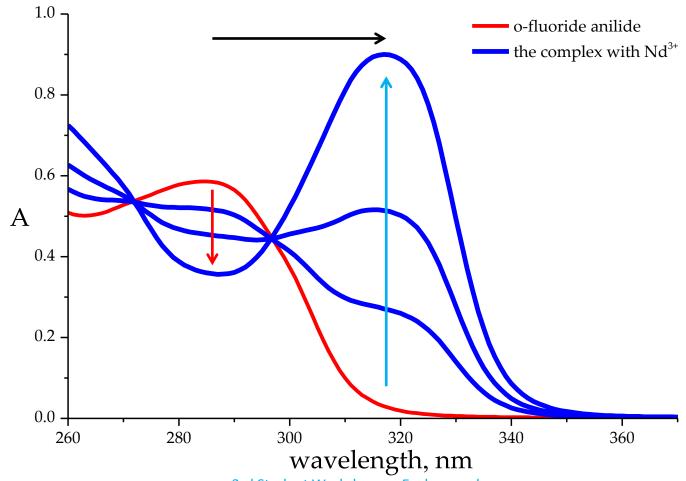


Prospective 2,2'-bipyridyl-6,6'-dicarboxamides

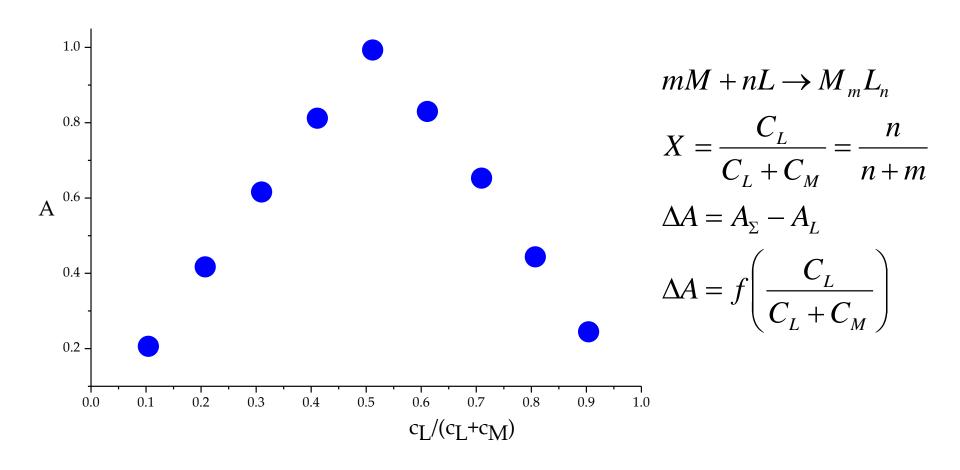
R =
$$o,p$$
-diMe (1) o,m -diMe (2); o,m,m -triMe (3); p -tret-Bu (4); o -F (5); m -F (6); p -F (7); o -Br (8); H (9^[1])

[1] European Journal of Inorganic Chemistry, 2014, 2219–2229

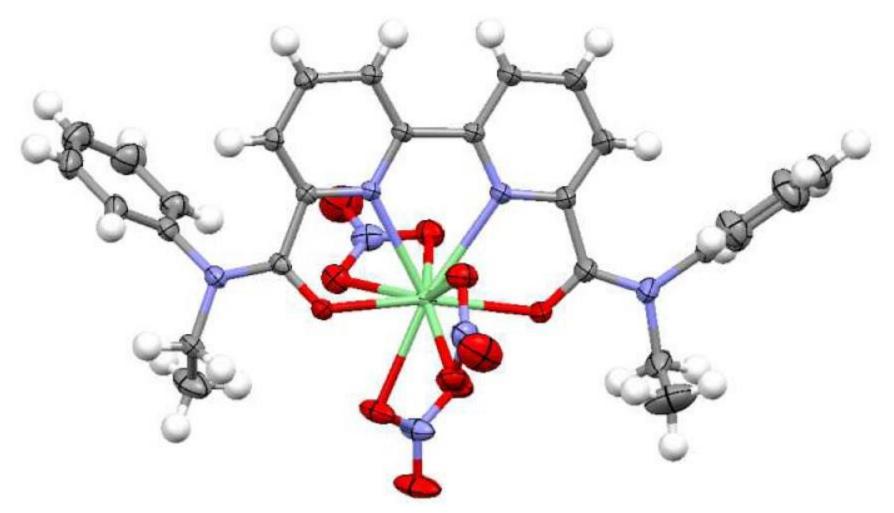
The absorption spectra of the o-fluoride anilide substituent and the complex with neodymium nitrate in dry acetonitrile



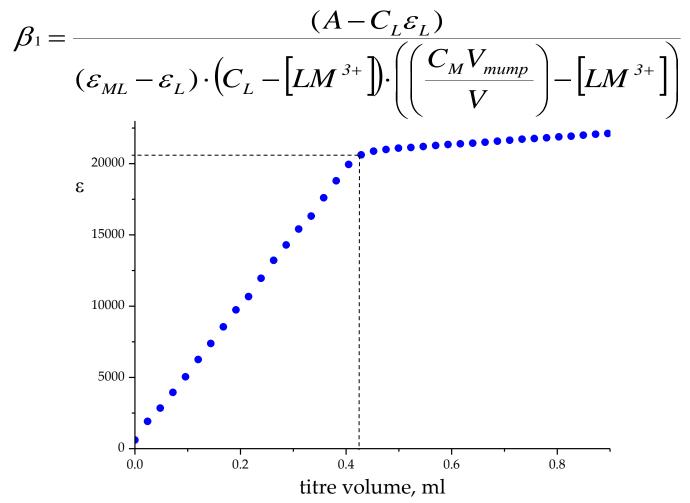
Job's plot for o-fluoride anilide with ion Tb³⁺ (in solution)



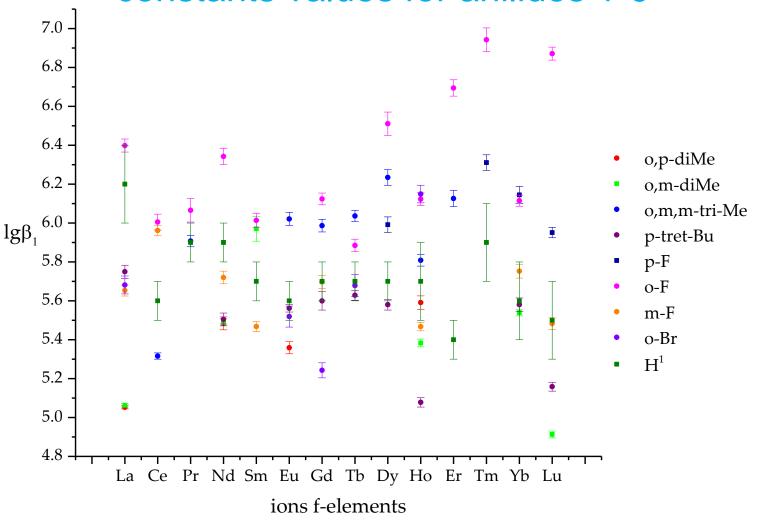
The structure of the ligand complex of unsubstituted anilide with gadolinium according to X-ray diffraction analysis (in solid state)



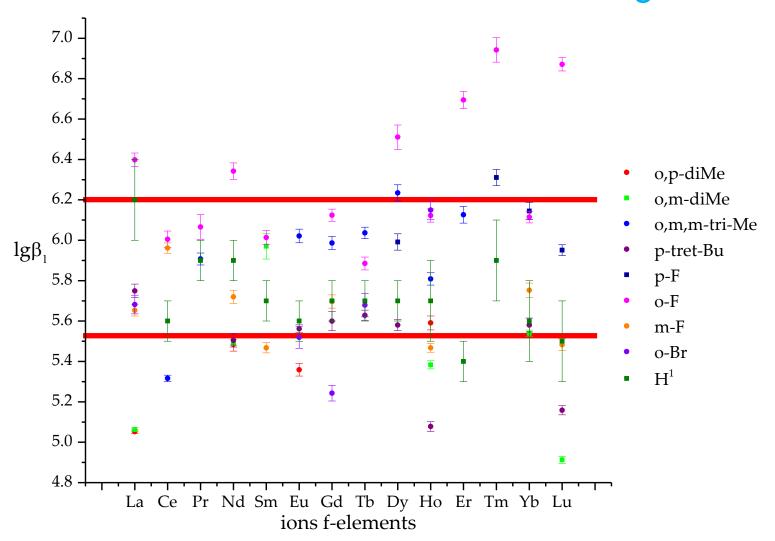
The dependence of the molar absorption coefficient on titre volume of metal for the ofluoride anilide with Nd3+ system



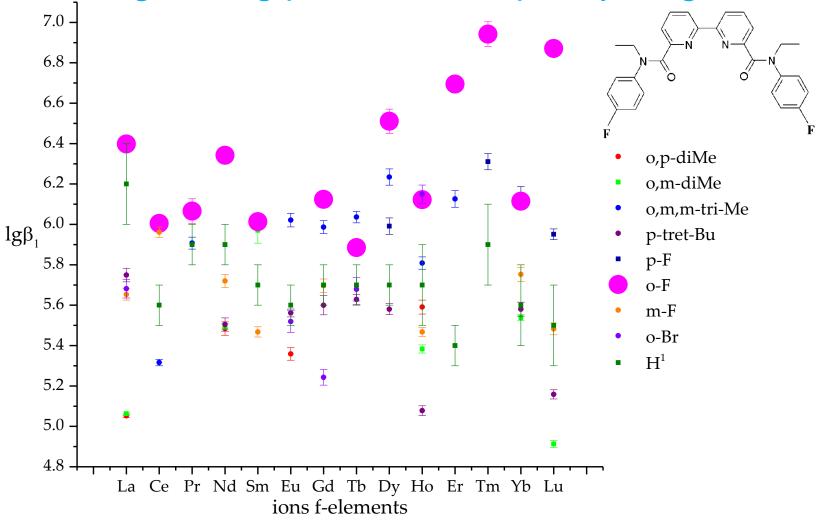
The lanthanide ion dependence of the stability constants values for anilides 1-9



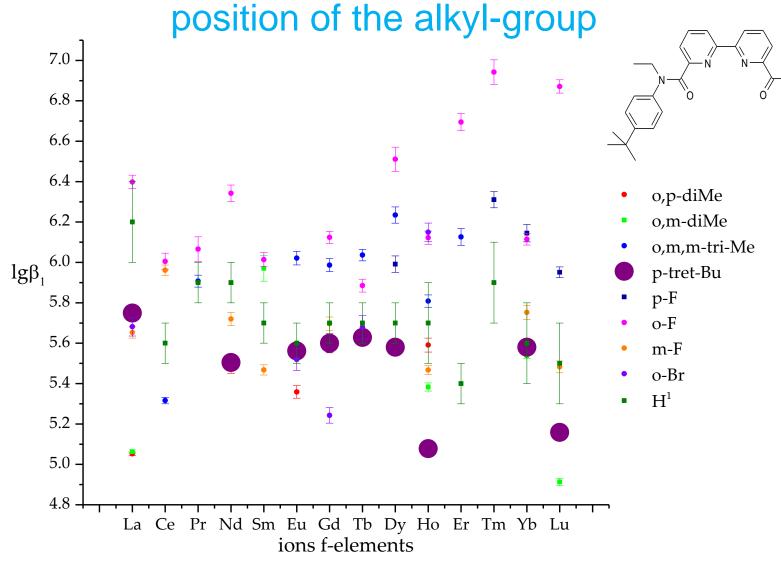
The lgβ₁ complexes of investigated anilides with lanthanide ions fall into 5,5-6,2 region



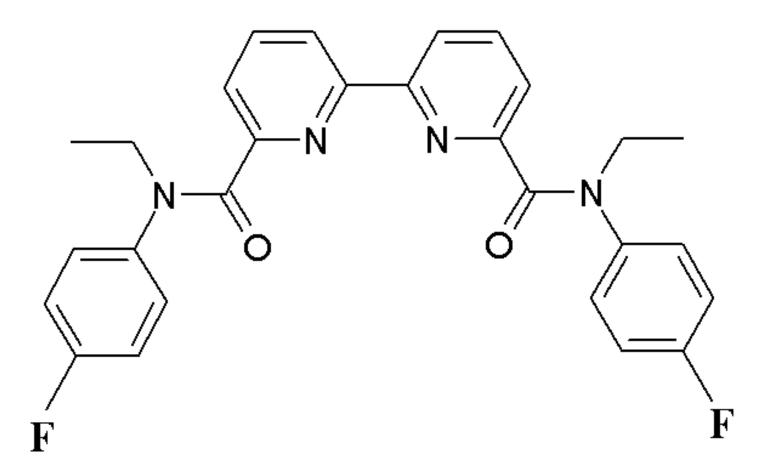
The fluorine in the o-position of the anilide stabilizes the complex due to conjugation of halogen long pairs with the phenyl ring



The long-chain substituent (t-Butyl) decrease the stability of the complexes irrespectively the



The o-fluorine-phenyl-amide could be serving as the most suitable candidate to clean industrial waste waters



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Thank you for your attention!

And to the



Organizing Committee

13 July, 2017, Kaliningrad